

ABSORPTION AND LUMINESCENCE PROPERTIES OF ACID AND SALT FORMS OF MONONUCLEOTIDES, THEIR COMPONENTS AND COMPLEXES WITH D-MANNITOL AT ROOM TEMPERATURE

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Today, medicines are being actively developed whose principle of action is based on the interaction of mononucleotides with biomolecules. At the same time, nucleotides in acidic form can differ greatly in their properties from salts, which was used in the development of the mRNA coronavirus vaccine [1]. Thus, studies of both this interaction and the mononucleotides themselves are very relevant. Our previous studies have shown that the addition of mannitol contributes to the antiviral effect of RNA molecules, unlike other sugars, amino acids, alkaloids, etc. Therefore, an interaction between mannitol and RNA molecules would be expected.

One of the most common areas of research is the use of spectroscopic methods that are non-destructive, relatively inexpensive, and provide important information about samples, such as electronic transition energies [2]. Heterocyclic molecules can luminesce because the energy difference between the ground and excited states of the electrons corresponds to the energy of the optical radiation that can be detected by spectroscopic methods.

However, spectroscopic studies of the radiation of aqueous solutions of nucleotides are usually performed at the boiling point of nitrogen or helium, when the sensitivity of most instruments allows for clear spectra [3–7]. In addition, most of the spectroscopic studies of mononucleotides to date have focused on their salts, which are widely available [8, 9]. However, this approach does not enable studying nucleotides in the state in which they are found in cells. Therefore, there is a need for spectroscopic studies of aqueous nucleotide solutions at room temperature.

Aim. The aim of this work was to analyze and compare spectral properties of aqueous nucleotide solutions and their mixes with D-mannitol in conditions close to biological systems.

Methods. We studied the absorption and luminescence (Ex and Em fluorescence and Em phosphorescence) of monoribonucleotides, their disodium salts, bases and nucleosides, and mixes with D-mannitol dissolved in water at room temperature. For these samples, we measured absorbance spectra using a Specord 210plus (Germany) instrument and fluorescence excitation and emission and phosphorescence spectra using Horiba Fluoro Max 4+ (USA) instruments.

Results and Discussion. We obtained absorption, excitation, and luminescence spectra of aqueous solutions 1 mg/ml of nucleotides, their components, and mixtures with mannitol (in ratio 1:4). We observed a change in the ratio between the peaks of the spectra of acidic and salt forms of nucleotides. The presence of mannitol practically did not affect the absorbance of most samples (except for cytosine). At the same time, the addition of D-mannitol could significantly change the appearance of the excitation, emission, and phosphorescence spectra of the samples. For example, the addition of D-mannitol changed the excitation spectrum of uridine monophosphate in its acidic form, as shown in the Figure.

It indicates the presence of low-affinity interactions between nucleotide molecules and mannitol and the possible formation of complexes. At the same time, the interaction with D-mannitol was different for nucleotides of acidic and salt forms.

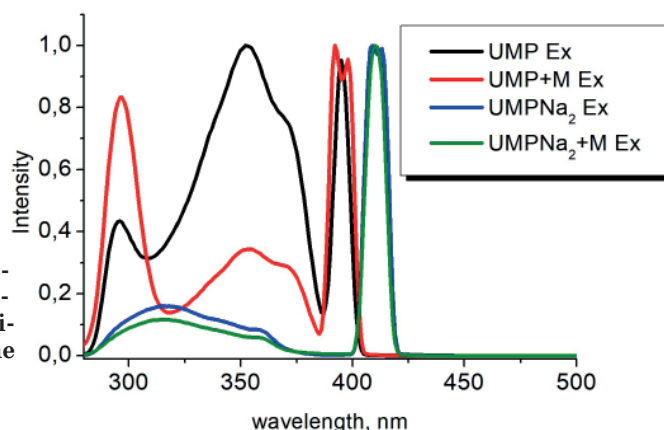


Fig. Normalized excitation spectra of uridine monophosphate in acidic form (UMP), uridine monophosphate in acidic form with D-mannitol (UMP+M), uridine monophosphate in salt form (UMPNa₂), uridine monophosphate in salt form with D-mannitol (UMPNa₂+M)

Conclusions. Thus, our observations confirm that nucleotides, nucleosides, and nucleic acid bases exhibit luminescence at room temperature, which may be useful information for further research in this area. A comparative analysis of the spectra showed possible interactions between nucleotide molecules and mannitol as well.

Key words: nucleic acids, luminescence, mannitol.

Authors' contribution. Conceptualization, Z.T.; methodology, R.N.; software, R.N.; validation, R.N.; formal analysis, R.N.; investigation, M.D. and R.N.; resources, Z.T.; data curation, R.N.; writing—original draft preparation, M.D. and R.N.; writing—review and editing, M.D., R.N. and Z.T.; visualization, M.D.; supervision, Z.T.; project administration, Z.T.

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